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Zyzzyanone A, a novel pyrrolo[3,2-f lindole alkaloid from the Australian marine sponge Zyzzya fuliginosa

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Abstract—A new dipyrroloquinone, zyzzyanone A 1, having a pyrrolo[3,2-f]indole-4,8(1H,7H)-dione skeleton, was isolated from the Australian marine sponge Zyzzya fuliginosa, along with the known pyrroloquinoline alkaloids, makaluvamines C, E, G, H, and L, and damirones A and B. The structure of 1 was determined by spectroscopic data. Zyzzyanone A 1 shows moderate cytotoxic activity against mouse Ehrlich carcinoma cells (IC₅₀ 25 µg/mL), inhibits the cell division of fertilized sea urchin eggs at a concentration of 25 µg/mL, and exhibits UV-A and UV-B absorbing activity.

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Marine sponges of the genera Latrunculia, Batzella, Prianos, Zyzzya are a rich source of alkaloids bearing a pyrrolo[4,3,2-de]quinoline skeleton. This series of alkaloids comprises of about 60 metabolites: discorhabdins, epinardins, batzellines, isobatzellines, makaluvamines, damirones, veiutamine.¹ Several structurally related alkaloids, for example, wakayin, isolated from the ascidian Clavelina sp.² and tsitsikammamines A and B, isolated from a South African latrunculid sponge,³ have a dipyrrolo[4,3,2-de:2',3'-h]quinoline core. Pyrroloquinoline alkaloids have shown a variety of biological activities such as inhibition of topoisomerase I⁴ and II,⁵ cytotoxicity against different tumor cell lines,^{5,6} inhibition of the phosphatase activity of calcineurin and peptidase activity of CPP32,7 antifungal,4 and antimicrobial8 activities.

In the course of our search for biologically active metabolites from marine sponges we have investigated the aqueous EtOH extract of the Australian marine sponge Zyzzya fuliginosa (order Poecilosclerida).[†] We have isolated a new dipyrroloquinone, named zyzzyanone A 1, together with the known makaluvamines C,⁵ E,⁵ G 5,⁴

H,9 and L,9 and damirones A and B.10 The known compounds were identified by comparison of their spectral data with published values. In this letter we describe the isolation and structural elucidation of the new compound 1.

The freeze-dried sponge (125 g) was extracted with 50% EtOH at room temperature. The extract was concentrated under reduced pressure to yield a dark red residue. This residue was subjected to column chromatography on a Polychrome-1 (powder Teflon) column with a solvent elution gradient from H₂O to EtOH. Dark red fractions eluted with 25-40% EtOH gave makaluvamines C^{5} H,⁹ and damirones A and B¹⁰ after chromatography on a Sephadex LH-20 column in CHCl₃-EtOH-TFA (4:1:0.1%). A brownish-green fraction eluted with 50% EtOH was chromatographed on a Sephadex LH-20 column in CHCl3-EtOH-TFA (3:1:0.1%) to yield makaluvamines E (0.004%),⁵ G (0.055%),⁴ L (0.008%),⁹ and among them compound 1 (0.006% on the dry weight of the sponge).

Zyzzyanone A 1 was obtained as a purple TFA salt, mp 300 °C (decomp.), which gave an orange color as a methanolic solution; UV–vis (MeOH) λ_{max} (log ε) 242 (4.26), 286 (3.90), 342 (3.59), 484 (3.44) nm; UV-vis (MeOH/ KOH) 243 (4.28), 286 (3.96), 346 (3.60), 505 (3.48) nm; IR (KBr) v_{max} 3600–2100 (with peaks at 3400, 3221), 1678, 1633, 1591, 1545, 1519, 1501, 1479, 1457, 1438,

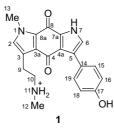
Keywords: Sponges; Dipyrroloquinones; Pyrrolo[3,2-f]indole; Pyrrolo[4,3,2-de]quinoline; Alkaloids.

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[†]Collected at Mid Island, Australian Great Barrier Reef, at a depth of 10m by hand using scuba.

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1402, 1382, 1270, 1204, 1182, 1134, 984, 909, 838, 798, 722 cm⁻¹; HRFABMS m/z 350.1483 (M⁺+H), C₂₀H₂₀N₃O₃ requires 350.1504.



Compound 1 showed a protonated molecular ion at m/z350.1483 by HRFABMS. The ¹³C NMR and DEPT spectra revealed 20 carbons and indicated the presence of 2 methyls, 2 methylenes, 6 methines, and 10 quaternary carbons (Table 1). On the basis of HRFABMS and ¹³C NMR data of 1, its molecular formula was established as C₂₀H₂₀N₃O₃. The UV and IR [1678, 1633 (C=O), 1591 (C=C)] spectra indicated the presence of a 1,4-benzoquinone moiety in 1. This was supported by the two quaternary carbon signals at δ 180.4 and 168.7 in the ¹³C NMR spectrum (Table 1). The IR spectrum of 1 also showed the band of a phenolic hydroxyl group (3400). The presence of a para-substituted phenol in 1 was evident from a pair of ortho-coupled doublets at δ 6.75 and δ 7.56, two protons each, and an exchangeable proton at δ 9.46 in the ¹H NMR spectrum. This was substantiated by COSY, DEPT, HSQC, and HMBC experiments.

The ¹H NMR spectrum of **1**, recorded in CD_3OD , in part resembled the spectra of pyrroloquinoline alkaloids. Aside from the signals of the *para*-substituted

Table 1. ${}^{13}C$ (125 MHz) and ${}^{1}H$ (500 MHz) NMR data of 1

phenol, the ¹H NMR spectrum contained two singlets of two *N*-methyl groups, a pair of triplets from two methylene groups, and two one-proton singlets in the aromatic region (Table 1). The values of coupling constants ¹J_{CH} (184 Hz) of these aromatic protons at δ 7.06 and δ 7.20, measured in gated decoupling experiments, showed that both belong to pyrrole rings. The ¹H NMR spectrum of **1**, recorded in DMSO-*d*₆ revealed three additional signals of exchangeable protons. Moreover, one of the signals in the aromatic region (δ 7.20) was now a doublet coupled with an exchangeable proton at δ 12.65, and a signal from one *N*-methyl group at 2.56 was a triplet coupled with a broad signal from two exchangeable protons at δ 8.51, as shown by a ¹H–¹H-COSY spectrum.

The presence of a 2,3,4-trisubstituted *N*-methyl-pyrrole ring in 1 was suggested by ¹H NMR [(δ 3.92 (s) and δ 7.06 (s)] and ¹³C NMR [35.8 (q), 129.5 (s), 125.4 (s), 119.6 (s), 130.3 (d, ${}^{1}J_{CH} = 184 \text{Hz})$] data (Table 1). The ${}^{1}\text{H}-{}^{1}\text{H}-\text{COSY}$ spectrum revealed a correlation of protons in a fragment from Me-13 to Me-12. The protons of Me-13 (δ 3.92) correlated with the proton at C-2 (δ 7.06), which itself exhibited long range correlation with protons at δ 2.99 of the mutually coupled methylene proton spin system. The protons at δ 3.10 of the mutually coupled methylene proton spin system correlated to a broad signal from two exchangeable protons at δ 8.51, which in turn coupled to the protons of Me-12 (δ 2.56). These correlation data suggested the protonated methylaminoethyl chain in 1, is attached to the N-methylpyrrole ring. This was supported by HMBC data. The HMBC correlations from the proton at δ 7.06 (H-2) to the carbons at δ 125.4, δ 129.5, and δ 119.6 supported the presence of the pyrrole ring. The carbon at δ 129.5 correlated to the N-methyl protons at δ 3.92 and to

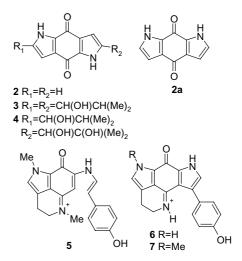
No.	CD ₃ OD				
	δC	$\delta H(J)$	δC	$\delta \mathbf{H}\left(J ight)$	HMBC
2	131.9	6.80 s	130.3 (CH)	7.06 s	3, 3a, 8a, 9, 13
3	121.4		119.6 (C)		
3a	127.9		125.4 (C)		
4	183.3		180.4 (C)		
4a	123.6		121.4 (C)		
5	129.4		126.6 (C)		
6	125.2	7.01 s	123.9 (CH)	7.20 d (2.5)	4a, 5, 7a, 14
7a	135.6		133.3 (C)		
8	170.9		168.7 (C)		
8a	132.3		129.5 (C)		
9	24.3	3.01 t (7.1)	22.0 (CH ₂)	2.99 t (7.4)	2, 3, 3a, 10
10	51.2	3.19 t (7.1)	48.1 (CH ₂)	3.10 m	3, 9, 12
12	34.3	2.67 s	32.6 (CH ₃)	2.56 t (5.5)	10
13	37.1	3.89 s	35.8 (CH ₃)	3.92 s	2, 8a
14	126.6		124.0 (C)		
15	131.7	7.54 d (8.7)	129.8 (CH)	7.56 d (8.7)	5, 17, 19
16	116.2	6.77 d (8.7)	114.6 (CH)	6.75 d (8.7)	14, 17, 18
17	158.6		156.7 (C)		
18	116.2	6.77 d (8.7)	114.6 (CH)	6.75 d (8.7)	14, 16, 17
19	131.7	7.54 d (8.7)	129.8 (CH)	7.56 d (8.7)	5, 15, 17
NH-7				12.65 d (2.5)	4a, 5, 7a
$N^{+}H_{2}-11$				8.51 m	
OH				9.46 b s	

the proton at δ 7.06, allowing assignment of this carbon as C-8a. The proton at δ 7.06 (H-2) and the protons at δ 2.99 (H₂-9) correlated to the carbon at δ 125.4 indicating it to be C-3a. The HMBC correlations from the proton at δ 7.06 (H-2) to the carbon at δ 22.0 (C-9), and from the protons at δ 2.99 (H₂-9) to the carbons at δ 130.3 (C-2), δ 119.6 (C-3), and δ 125.4 (C-3a) pointed to the position of the methylaminoethyl chain being at C-3.

The presence of another 2,3,4-trisubstituted pyrrole ring was suggested by ¹H NMR [(δ 12.65 (d, J = 2.5 Hz) and δ 7.20 (d, J = 2.5 Hz)] and ¹³C NMR [133.3 (s), 121.4 (s), 126.6 (s), 123.9 (d, ¹ J_{CH} = 184 Hz)] data (Table 1). The HMBC correlations from the proton at δ 7.20 (H-6) and from the exchangeable proton at δ 12.65 (NH) to the carbons at δ 133.3, δ 121.4, and δ 126.6 supported the presence of the pyrrole ring. The position of the attachment of the *para*-substituted phenol to this pyrrole ring and assignment of C-5 were decided on the basis of HMBC data. Key HMBC correlations from the pyrrole proton at δ 7.20 (H-6) to the carbon atom at δ 124.0 (C-14) of the phenol ring and from the protons H-15(19) at δ 7.56 of the phenol ring to the carbon at δ 126.6 indicated the attachment of the phenol to this carbon (δ 126.6), and have allowed assignment of this carbon as C-5, as shown in 1.

The ¹H and ¹³C NMR spectra and the unsaturation requirements of the molecular formula indicated 1 to be tetracyclic. Indeed, the presence of a tetrasubstituted 1,4-benzoquinone unit, two 2,3,4-trisubstituted pyrrole rings, and the *para*-substituted phenol in 1 were substantiated by COSY, DEPT, HSQC, and HMBC experiments. Unfortunately, the connectivity between a tetrasubstituted 1,4-benzoquinone unit and two 2,3,4-trisubstituted pyrrole rings was not fully clarified on the basis of the HMBC correlations. The data obtained allowed for two skeletons, **2** and **2a**, as the possible tricyclic backbone of **1**.

Two dipyrroloquinones, terreusinone **3**, and terreusinol **4**, having a skeleton **2**, have been recently isolated as the products of the marine-derived fungus *Aspergillus terreus*.^{11,12} Dipyrroloquinones **2** and **2a** were known previously as the products of isomerization of dipyrrolo[1,2-*a*:1',2'-*d*]pyrazine-5,10-dione during a flash vacuum pyrolysis at 850–900 °C.¹³ When we compared the ¹³C NMR data of the 1,4-benzoquinone units in com-



pounds with skeletons 2 and 2a (Table 2), we noted that significant differences in ¹³C chemical shifts occurred at C-4, C-4a, C-7a, and C-8. The presence of substituents on pyrrole rings have no significant influence on the chemical shifts of the carbonyl carbons in compounds 1, 3, and 4 in comparison with the parent dipyrroloquinones 2 and 2a. Thus, the chemical shift at δ 180.4 is a diagnostic C-4 chemical shift for the presence of the skeleton 2a in compound 1. The structure of zyzzyanone A was thus determined to be pyrrolo[3,2-*f*]indole-4,8(1*H*,7*H*)-dione.

To confirm the structure of zyzzyanone A 1, we prepared it from makaluvamine G 5. Treatment of an EtOH solution of 5 with aqueous NH₃ at room temperature for 10h gave a mixture of products. Chromatography of the reaction mixture on a Sephadex LH-20 column in CHCl₃–EtOH–TFA (3:1:0.1%) yielded damirone A¹⁰ (14.5%) and a cyclized product 1 (32%), the spectral and physical data of which were identical with those for zyzzyanone A. Thus the structure 1 was confirmed.

The marine sponge Z. fuliginosa has been investigated for makaluvamines and damirones,^{5,9} veiutamine,¹⁴ batzellines, and isobatzellines,¹⁵ discrhabdins,¹⁶ makaluvic acids,¹⁷ and modified purines.¹⁸ To our knowledge, compound **1** is the first example of a pyrrolo[3,2-f]indole-4,8(1H,7H)-dione alkaloid isolated from a marine

Table 2. ¹³C NMR data for 1,4-benzoquinonoe units of 1–4 (DMSO-*d*₆)

No.	δC						
	1	2a ^a	2 ^a	3 ^b	4 ^c		
3a	125.4	127.3	125.7	126.0	127.7		
4	180.4	179.3	174.0	174.0	173.9		
4a	121.4	127.3	132.9	131.3	130.9		
7a	133.3	131.9	125.7	126.0	125.7		
8	168.7	168.3	174.0	174.0	174.0		
8a	129.5	131.9	132.9	131.3	131.3		

^a Data reported by Qiao et al.¹³

^b Data reported by Lee et al.¹¹

^c Data reported by Li et al.¹²

sponge. The tricyclic dipyrroloquinone skeleton of zyzzyanone A 1 can be considered as a *seco*-derivative of a tetracyclic dipyrroloquinoline core of tsitsikammamines A 6 and B 7, isolated from the South African latrunculid sponge.³ Zyzzyanone A could arise from makaluvamine G 5 by intramolecular cyclization at the benzylic position with a concomitant hydrolysis of an imino bond in a hypothetical analogue of tsitsikammamines. Thus, zyzzyanone A can be seen to have a plausible interrelationship with the makaluvamines and with the tsitsikammamines.

Zyzzyanone A 1 showed moderate cytotoxic activity against mouse Ehrlich carcinoma cells ($IC_{50} 25 \mu g/mL$), inhibited the cell division of fertilized sea urchin eggs at concentration of $25 \mu g/mL$, and exhibited UV-A and UV-B absorbing activity. An investigation of UV-protecting activity is in progress.

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